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Ionic Interactions In Natural And Synthetic Macromolecules

The interaction of ions with less polar sites on the macromolecule (e.g. amide bonds), and the origin of the lyotropic effects (focusing on binding versus condensation) will also be extensively addressed.

Ionic Interactions in Natural and Synthetic Macromolecules

Description This book is a comprehensive study of the subject of ionic interactions in macromolecules. The first parts of the book review and analyze the conventional treatments of fixed charges (e.g. in polyelectrolytes and polyampholytes), including screening and condensation by mobile ions.

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The interactions of the ionic components of natural waters have been examined using the ion pairing and specific interaction models. The present paper reviews the recent developments made in using these models to estimate the activity and speciation of divalent metals in natural waters as a function of ionic strength.

Ionic interactions of divalent metals in natural waters

His research has had significant impact in the areas of the rheological properties of natural and synthetic polymers, fibrous proteins, interactions between salts and macromolecules, liquid crystals, and supramolecular assemblies.

Wiley: Ionic Interactions in Natural and Synthetic

Ionic interactions may take place between cationic groups, for example, in between N-terminal amines like arginine and guanidine, and anionic groups like carboxylic acid in aspartic and glutamic acid. As discussed earlier, these ionic interactions, and thereby electrostatic energy, are inversely proportional to the dielectric constant of the medium.

Ionic Interaction - an overview | ScienceDirect Topics

The interactions of the new and old components appeared automatically in the differential equations without further physical argument. In this way, complex systems could be studied without ad hoc physical study of interactions, if the original and composite systems involved the same physics described by similar Hamiltonians.

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